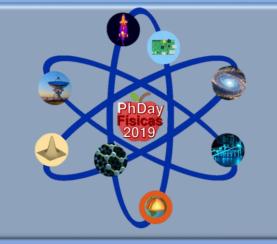


## Synthesis and characterization of doped $SnO_2$ and $TiO_2$ nanoparticles and hybrid composites for technological applications



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### INTRODUCTION

Wide bandgap semiconducting oxides, such as SnO<sub>2</sub> and TiO<sub>2</sub>, in form of nanoparticles have demonstrated potential applicability in numerous fields of technology such as optoelectronic devices, catalysis, gas sensing or energy storage. A higher control of the dimensions and doping of these nanoparticles can lead to enhanced performance.

Hybrid composites are emerging as low-cost materials with reported applications in photovoltaic and energy storage devices, among others [1]. In addition to the characteristic properties of the organic and inorganic compounds, these materials can exhibit new synergetic properties.

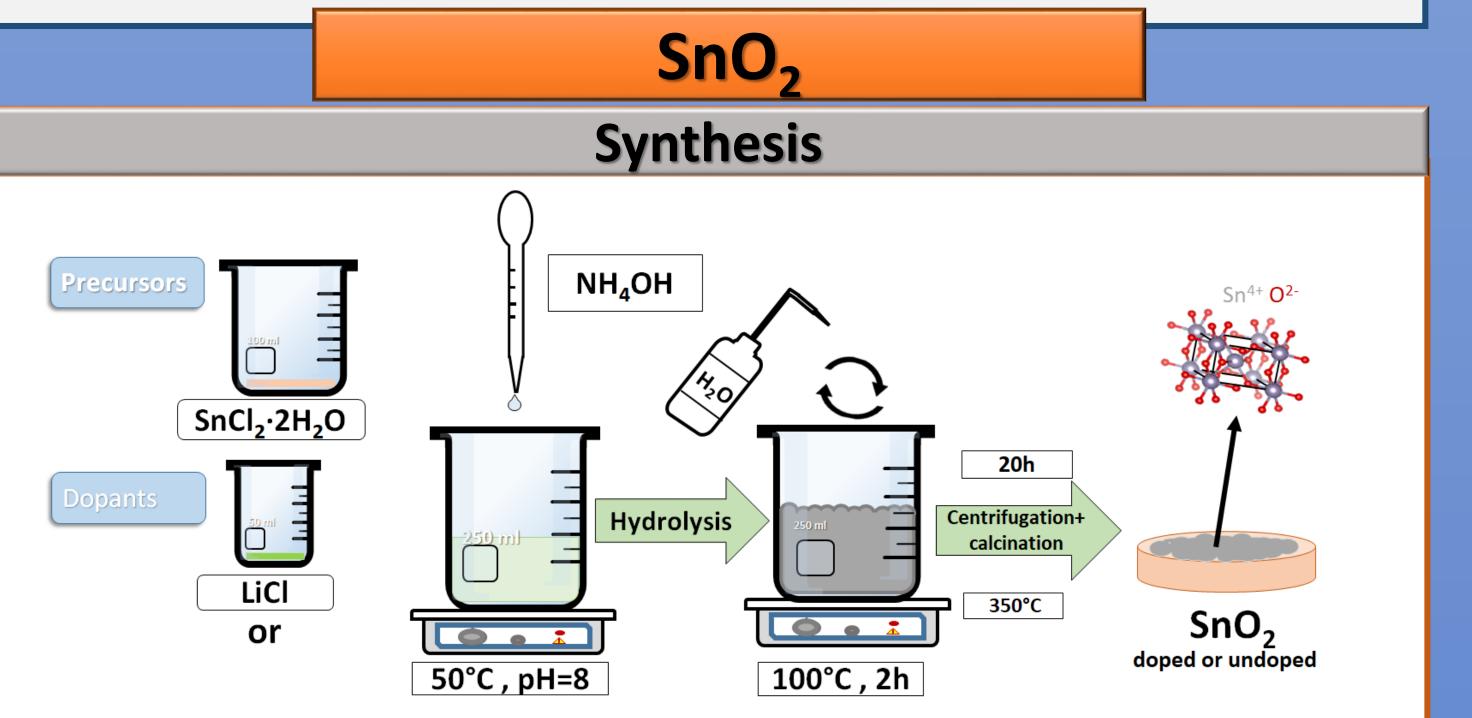
#### AIMS

- Synthesis via co-precipitation method based on hydrolysis of both anatase TiO<sub>2</sub> and rutile SnO<sub>2</sub> nanoparticles doped with Li or Ni, with variable cationic concentrations.
- Structural, compositional and luminescent characterization of the obtained nanoparticles.
- Fabrication of a composite via spin-coating technique combining the nanoparticles with PEDOT: PSS and measurement of **photovoltaic** properties.

TiO<sub>2</sub>

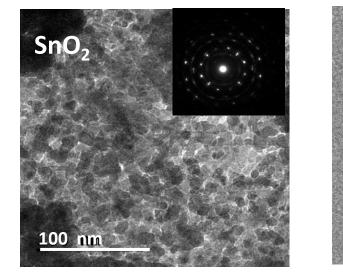
**Synthesis** 

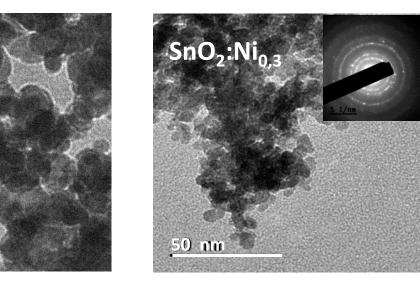
• Assembly and measurement of different anodes for ion-Li batteries.



## Characterization

#### **Transmission Electron Microscopy (TEM)**





The presence of dopants modifies the SnO<sub>2</sub>

PL spectra show common SnO<sub>2</sub> emissions [2]

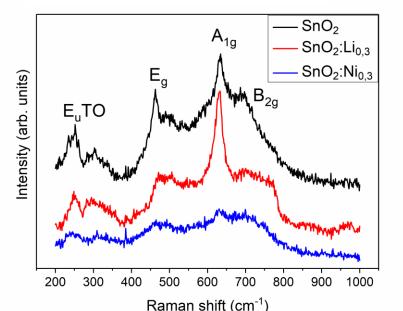
decrease the luminescence ("killer effect").

vibrational modes (mainly E<sub>g</sub> and A<sub>1g</sub>

modes).

TEM analysis confirms the presence of SnO<sub>2</sub> nanoparticles with dimensions 4 – 20 nm, in accordance with XRD analysis. SnO<sub>2</sub>:Ni<sub>0.3</sub> nanoparticles show the lowest dimensions.

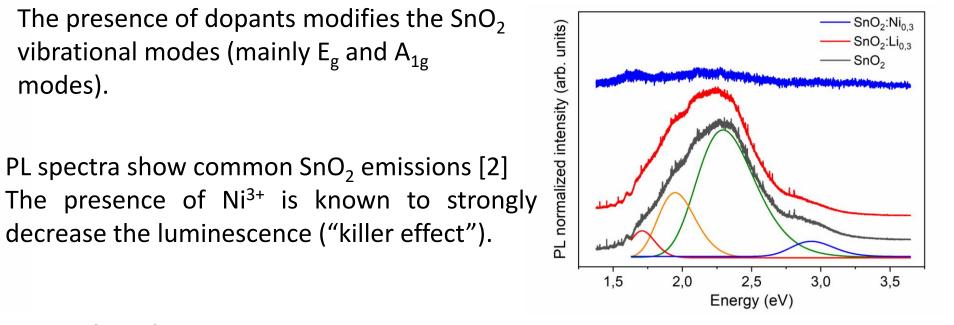
#### **Raman Spectroscopy**

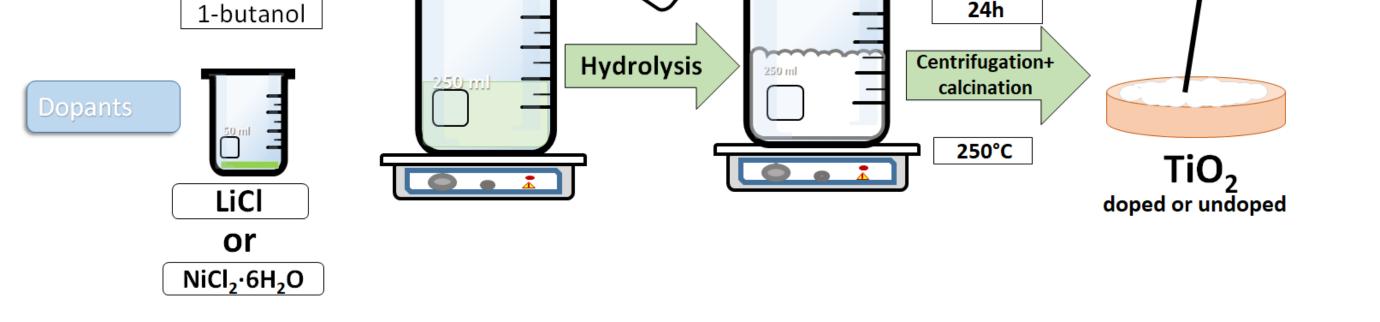


#### X-ray Photoelectron Spectroscopy (XPS)

<u> </u>	17%		 $\sim$	
÷≝		(s)	l ts	
⊆		Only Sn4+ in both		
		$(Jn)V$ $Nn^{-1}$ in noth $c \mid n$		•



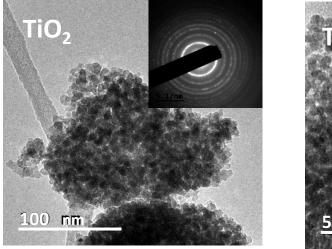




## Characterization

#### Transmission Electron Microscopy (TEM)

Ti(OBut)₄

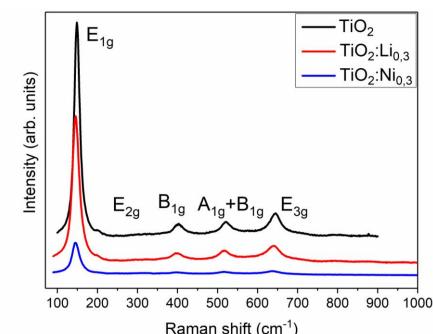


# 50 nm

#### **Raman Spectroscopy**

s)

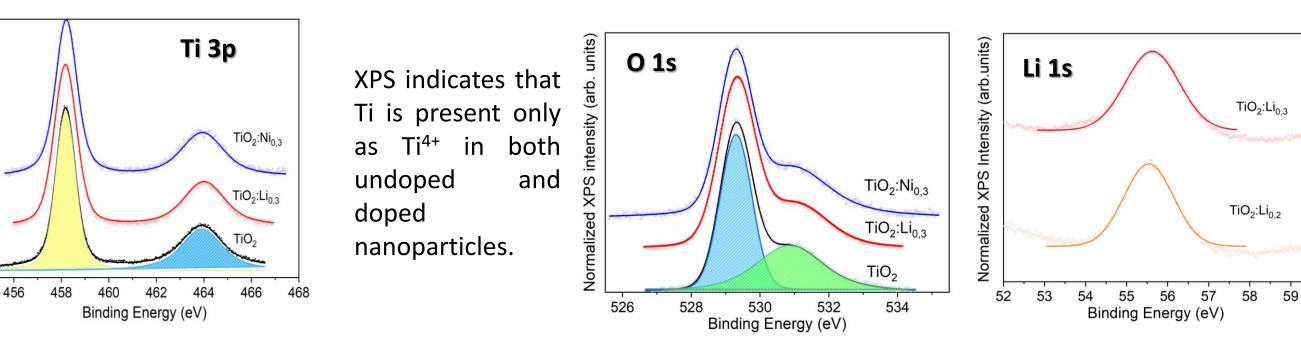
Precurso



Small variations have been observed in the anatase TiO<sub>2</sub> vibrational modes due to the presence of dopants.

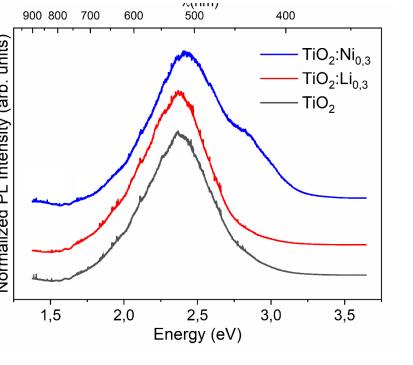
A broad emission centred in 2,36 eV related to oxygen vacancies dominates the PL spectra. Ni doped nanoparticles also exhibit an emission at 2,78 eV [3]

#### X-ray Photoelectron Spectroscopy (XPS)



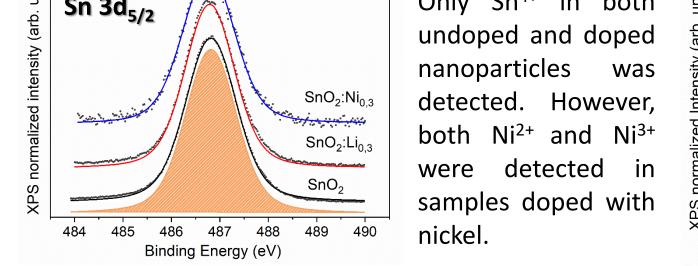
patterns confirm the SAED presence of anatase TiO<sub>2</sub> with dimensions 4 – 15 nm. nanoparticles show Doped reduced dimensions, according to TEM analysis.

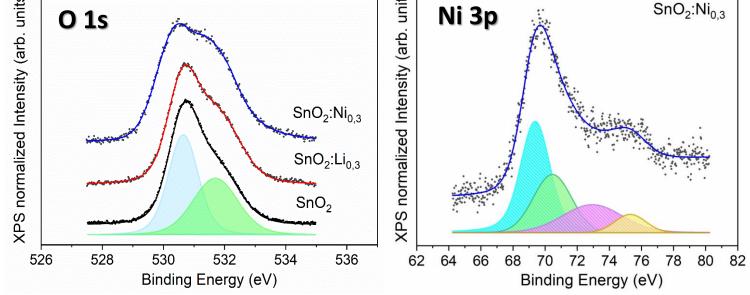




TiO<sub>2</sub>:Li<sub>0,3</sub>

TiO<sub>2</sub>:Li<sub>0,2</sub>





#### **Compositional analysis**

EDS	EDS		ICP-OES		
dopant conc.	% at.	dopant conc.	% at.		
Ni 0,3	3,9±0,3	Li 0,3	0,95±0,02		
Ni 0,2	0,5±0,1	Li 0,2	0,06±0,03		

The amount of Ni in the nanoparticles is below 4 % at., according to EDS measurements. Li is incorporated in the SnO<sub>2</sub> lattice in concentrations below 1 % at, as detected by ICP-OES.

**Compositional analysis** 

EDS	EDS		ICP-OES		
dopant conc.	% at.	dopant conc.	% at.		
Ni 0,3	1,5±0,2	Li 0,3	0,65±0,02		
Ni 0,2	1,4±0,1	Li 0,2	0,50±0,02		

Concentrations below 1,5 at. % have been estimated for Ni and Li, respectively. Slight variations in the at. % have been observed for both dopants.

## BATTERIES

#### **FABRICATION**

Anodes Li- ion batteries were fabricated combining SnO<sub>2</sub> with active material and binder in different concentrations, for wich carbon black and PAA were selected.

#### **Electric measurements**

	2509 -
Spouli shows an increase of conductivity and corrier	( <b>b</b> ) 2508 -
SnO <sub>2</sub> :Li <sub>0,2</sub> shows an increase of conductivity and carrier	UN 2507 -
density.	E 2506 -
However adding nickel decrease both parameters,	> 2505 -
specially for higher concentrations.	- <b>- - -</b>
	1750 - d 1700 -
Carrier concentraton _ <sub>10</sub> <sup>20</sup>	<u>ල</u> 1650 –
	0 1600 -

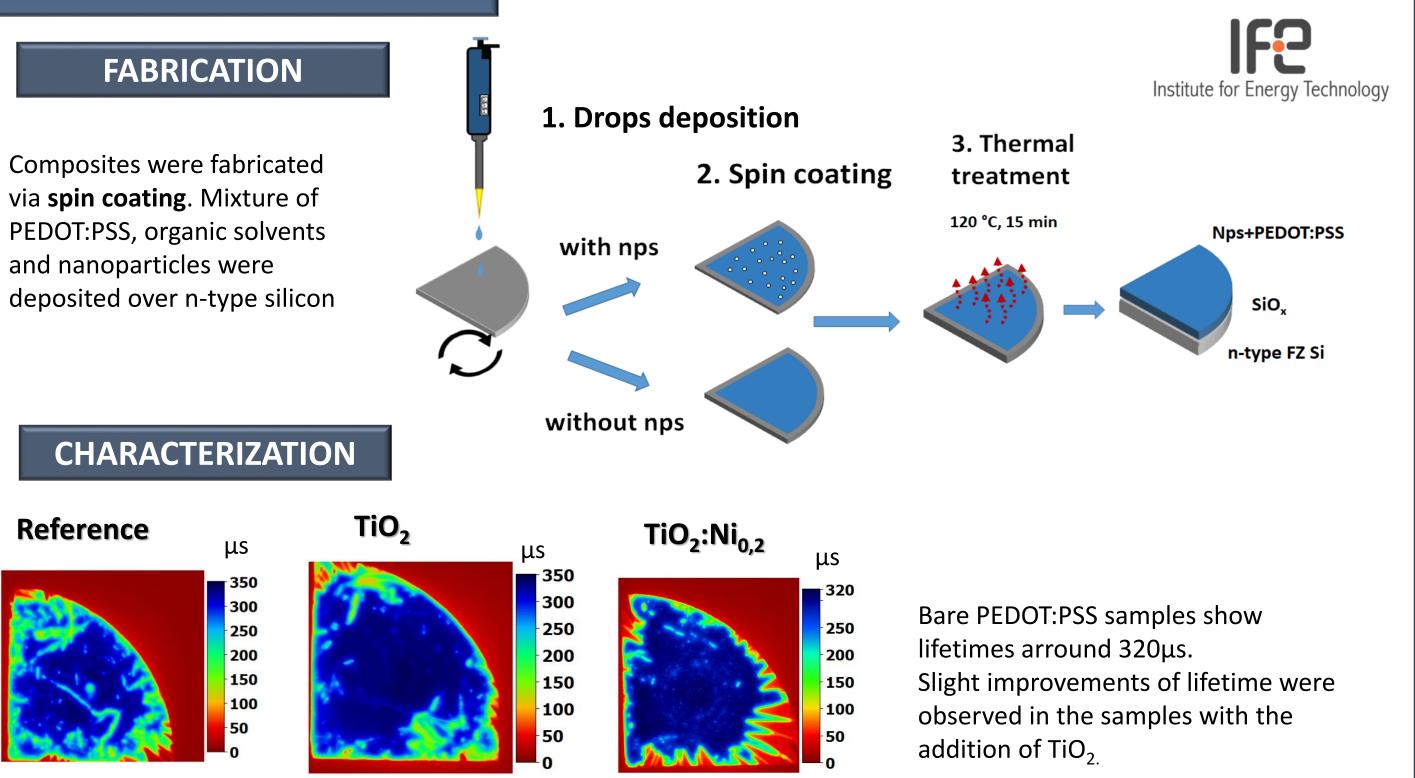
CHARACTERIZATION

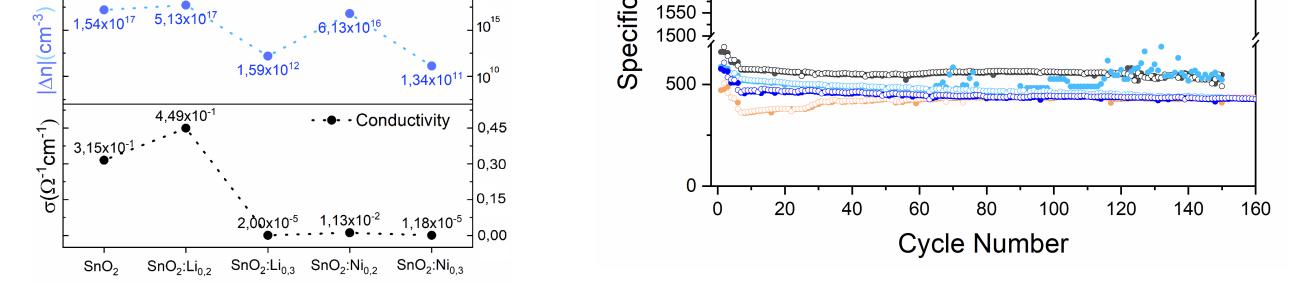


After 100 cycles, SnO<sub>2</sub> sample shows a specific capacity of 562mAh/g. Doped ones show great stability over time.

	0.0 Observes
•	SnO <sub>2</sub> Charge
0	SnO <sub>2</sub> Discharge
•	SnO <sub>2</sub> :Li <sub>0,2</sub> Charge
0	SnO <sub>2</sub> :Li <sub>0,2</sub> Discharge
•	SnO <sub>2</sub> :Ni <sub>0,2</sub> Charge
0	SnO <sub>2</sub> :Ni <sub>0,2</sub> Discharge
•	SnO <sub>2</sub> :Ni <sub>0,3</sub> Charge
0	SnO <sub>2</sub> :Ni <sub>0,3</sub> Discharge

## HYBRID COMPOSITE





2510

#### REFERENCES

[1] M García-Tecedor, S. Zh Karazhanov, G.C. Vásquez, H. Haug, D. Maestre, A. Cremades, M. Taeño, J. Ramírez-Castellanos, J.M. González-Calbet, J. Piqueras, C.C. You, E.S. Marstein. *Nanotechnology*, **29**, 035401 (2018) [2] M. A. Peche-Herrero, D. Maestre, J. Ramírez-Castellanos, A. Cremades, J. Piqueras, J. M. González-Calbet, CrystEngComm, 16, 2969, (2014) [3] G. C. Vásquez, M. A. Peche-Herrero, D. Maestre, B. Alemán, J. Ramírez-Castellanos, A. Cremades, J. M. González-Calbet, J. Piqueras, J. *Mater. Chem. C* ,**2**, 10377–10385, (2014).

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#### CONCLUSIONS

- SnO<sub>2</sub> and TiO<sub>2</sub> nanoparticles, undoped and doped with Li or Ni, were synthesized using the hydrolysis method.
- Homogeneity in the dimensions and crystalline structure of the nanoparticles have been achieved.
- The dopant inclusion into the lattice does not modifies the chemical structure of the SnO<sub>2</sub> and TiO<sub>2</sub> nanoparticles but it does modify the Raman modes and luminescence.
- Anodes for ion-Li batteries were fabricated with SnO<sub>2</sub> as an active material, showing promising results on specific capacity for electrodes with SnO<sub>2</sub>
- Hybrid composites were fabricated by mixing PEDOT:PSS and TiO<sub>2</sub> showing good lifetime results for samples with  $TiO_2$  and  $TiO_2$ :Ni<sub>0.2</sub>