

Growth and characterization of micro- and nanostructures of ZnO co-doped with Erbium and Lithium



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Introduction

Why ZnO?

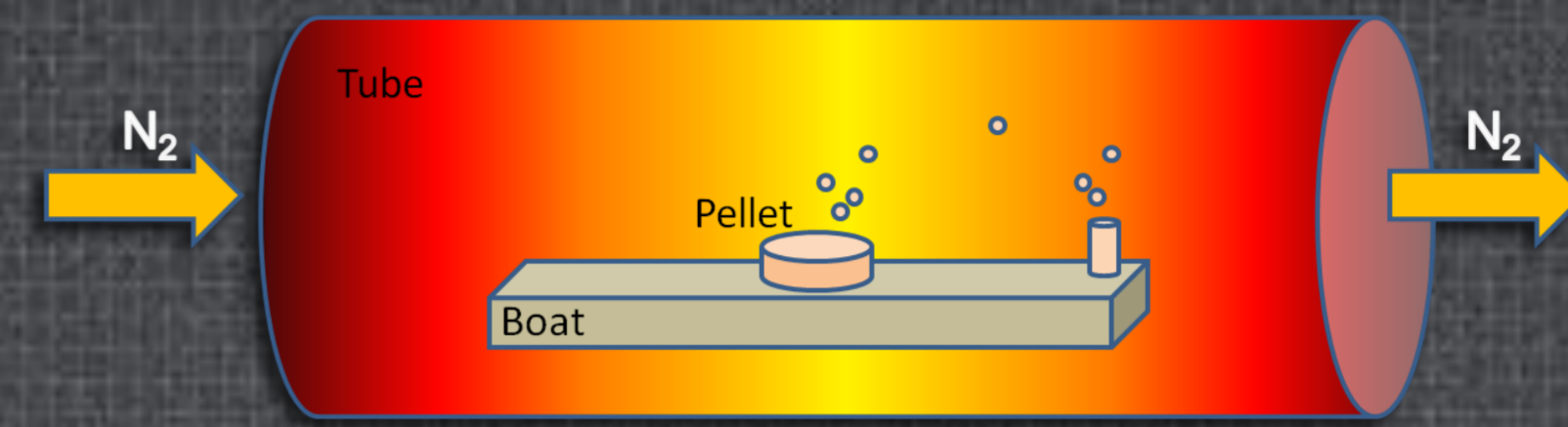
- Direct wide band gap semiconductor (3.37 eV)
- Large exciton binding energy (60 meV)
- Resistivity can be tailored between 10^{10} and $10^{-4} \Omega \cdot \text{cm}$
- Refraction index > 2
- Biocompatible, piezoelectric, non-ohmic behaviour... Excellent candidate for optoelectronic devices, gas sensing, piezotronics and Transparent Conductive Oxide (TCO)

Why Er and Li?

- Er { Intense intraionic emissions (1.54 μm widely used in optical communications)
 But at room temperature, the emission is strongly inhibited due to a non radiative transition
- Li { Ionic radii: $\text{Li}^+ \approx \text{Zn}^{2+} < \text{Tb}^{3+}$
 Charge balance: $\text{Li}^+ + \text{RE}^{+3} = \text{Zn}^{2+}$

Experimental techniques

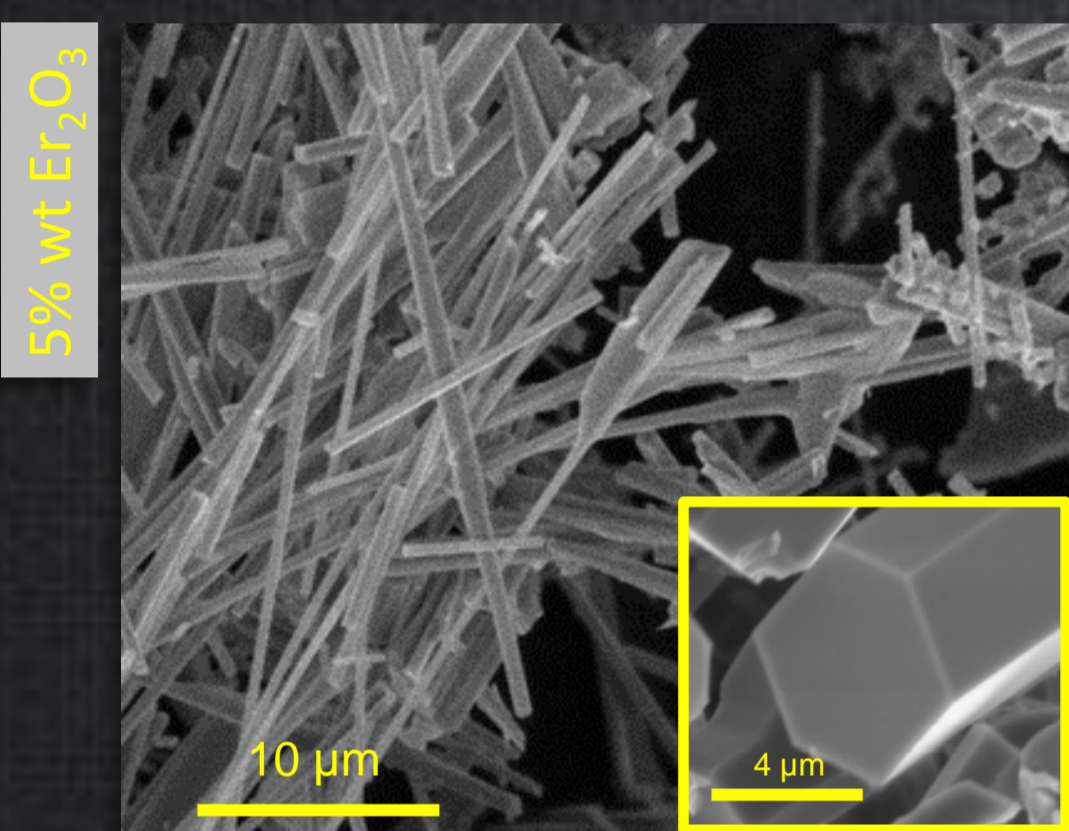
- $\text{ZnS} + 5\% \text{wt Er}_2\text{O}_3 + 1-10\% \text{wt Li}_2\text{O}$
- VS Method: 950°C , 10 h, 1.5 l/m N_2 flow



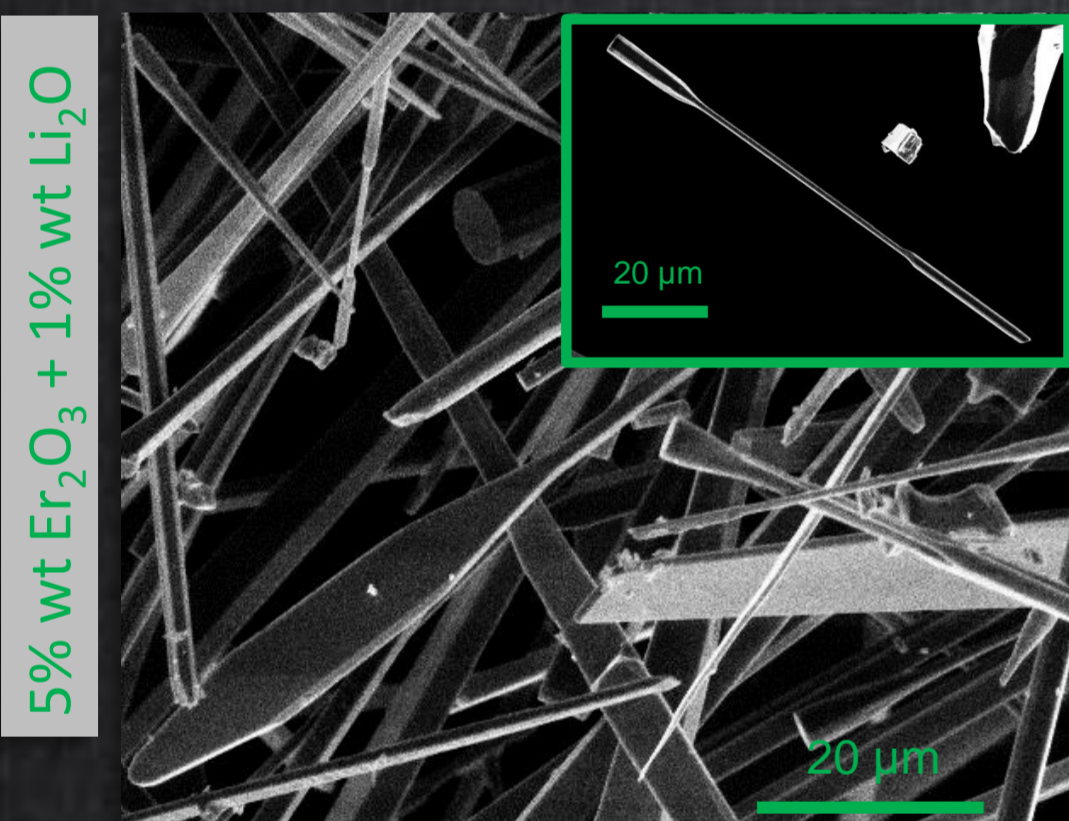
- Morphology: Scanning Electron Microscopy (SEM)
- Composition: X-Ray Microanalysis (EDS)
- Defect structure, optical properties: Cathodoluminescence (CL), Photoluminescence (PL), μ -Photoluminescence (μ -PL)

Morphology

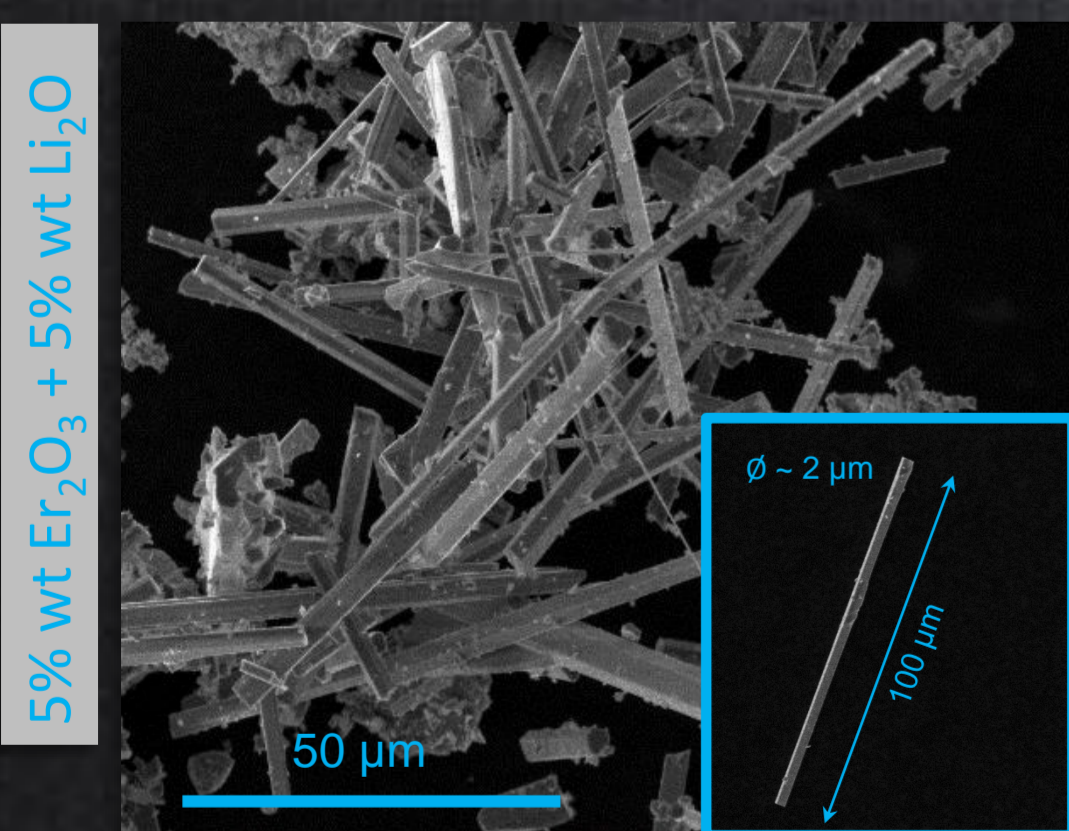
Amounts of erbium in a variable quantity are found in the different structures



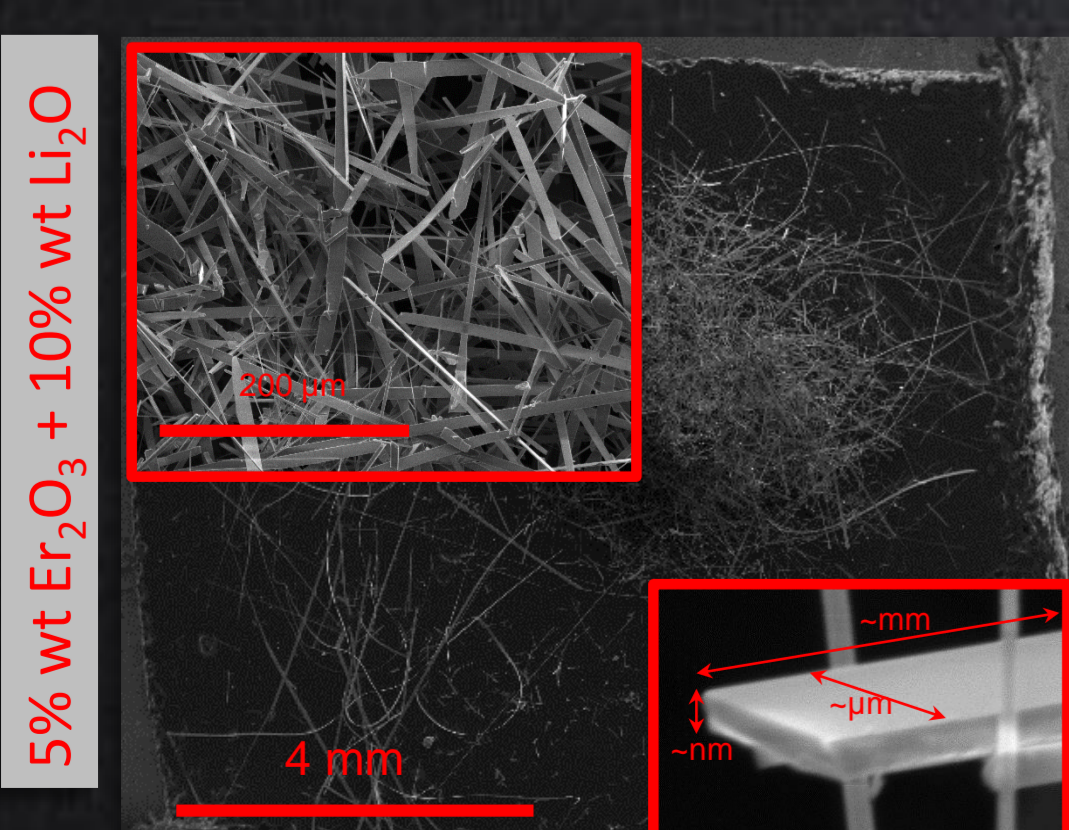
Furnace walls and boat. Rods tens of μm in length and 500 nm to 1 μm in diameter



Ribbons tens of μm in length and widths of hundreds of nanometers. Some screw and hexagonal prisms began to appear



Much more hexagonal rods of tens to hundreds μm in length and few μm in width

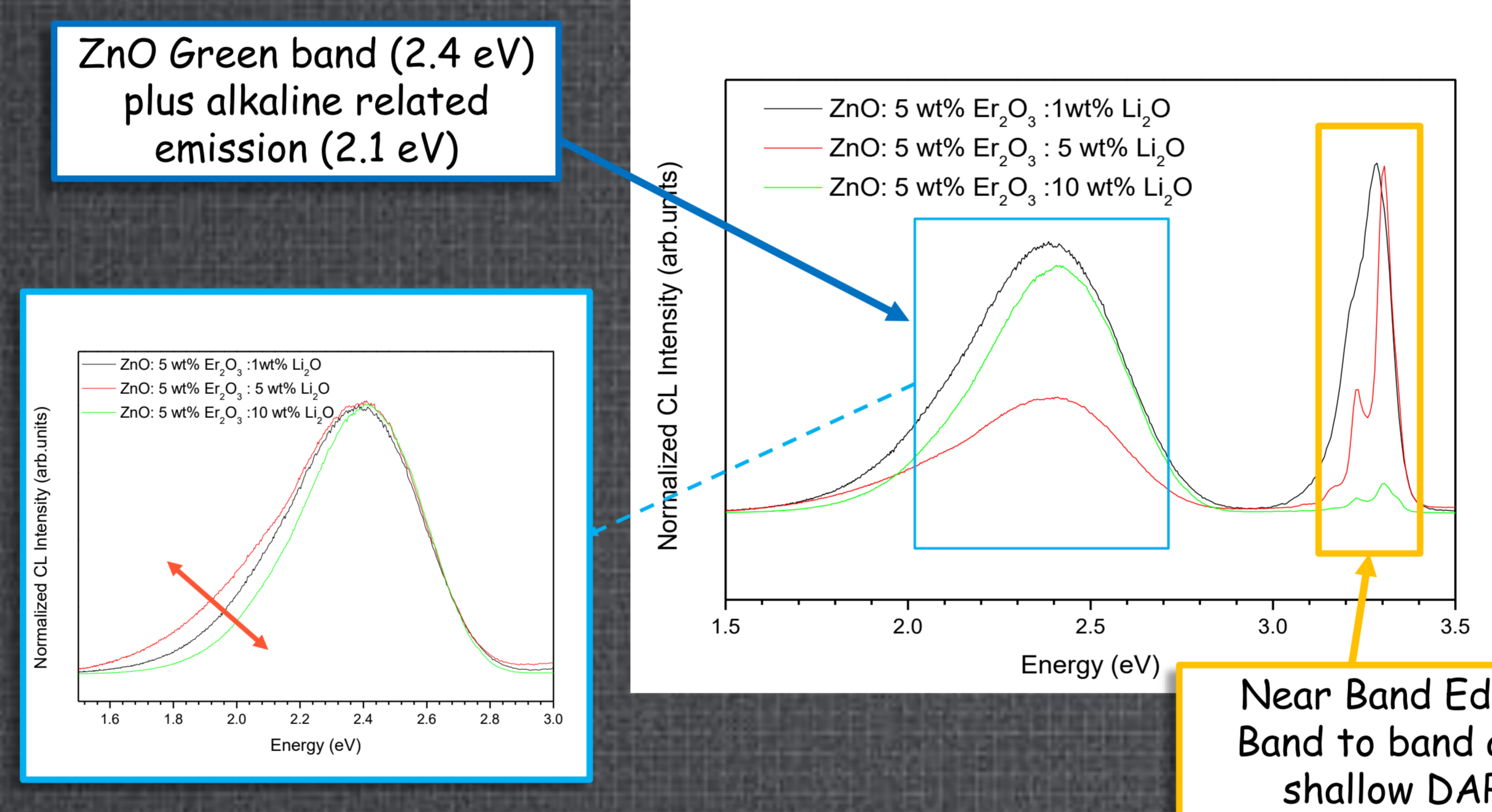


Ribbons, several millimeters in length, several microns width and hundreds of nanometers thick

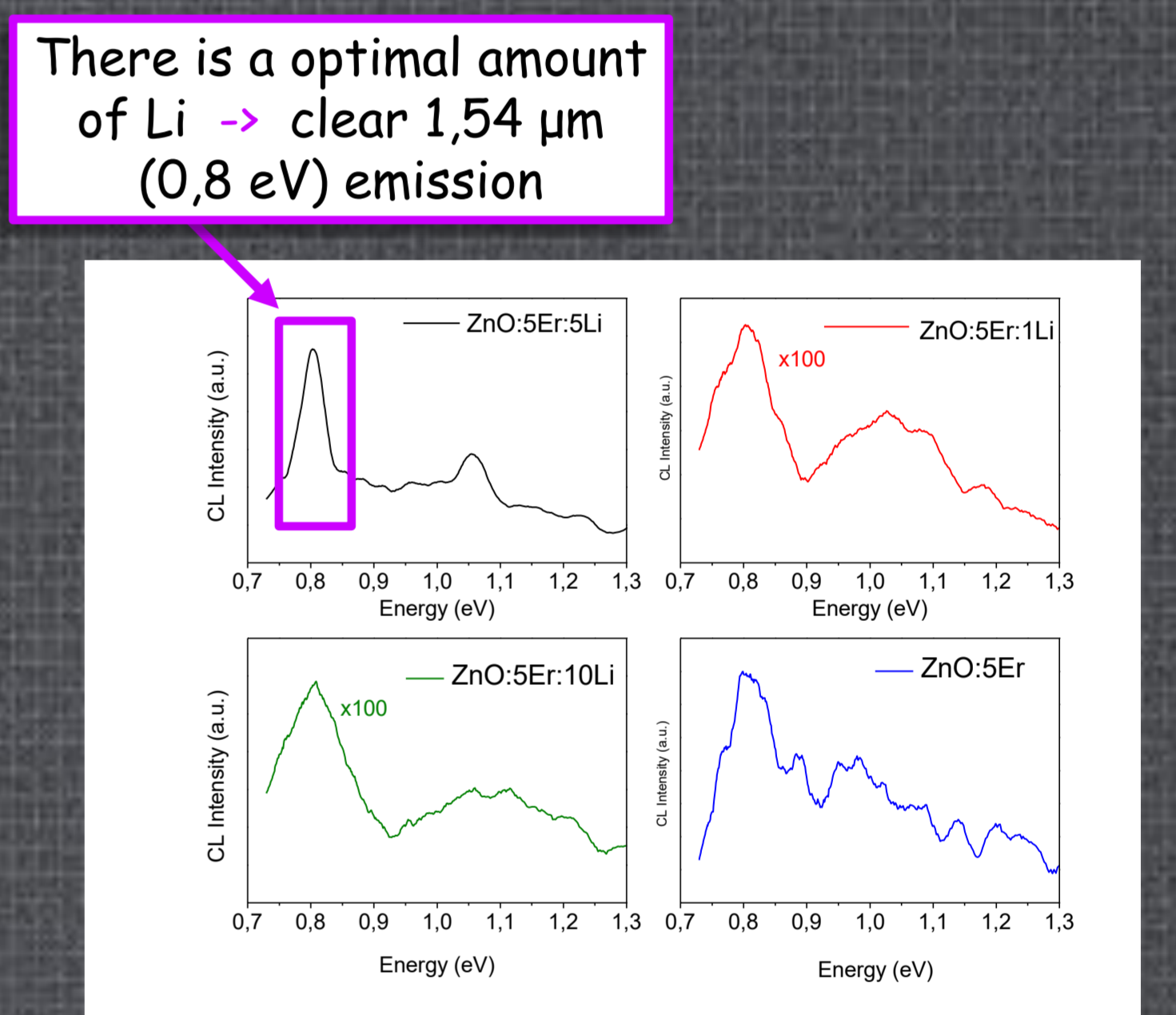
%Li₂O

Cathodoluminescence

UV-VIS CATHODOLUMINESCENT EMISSION

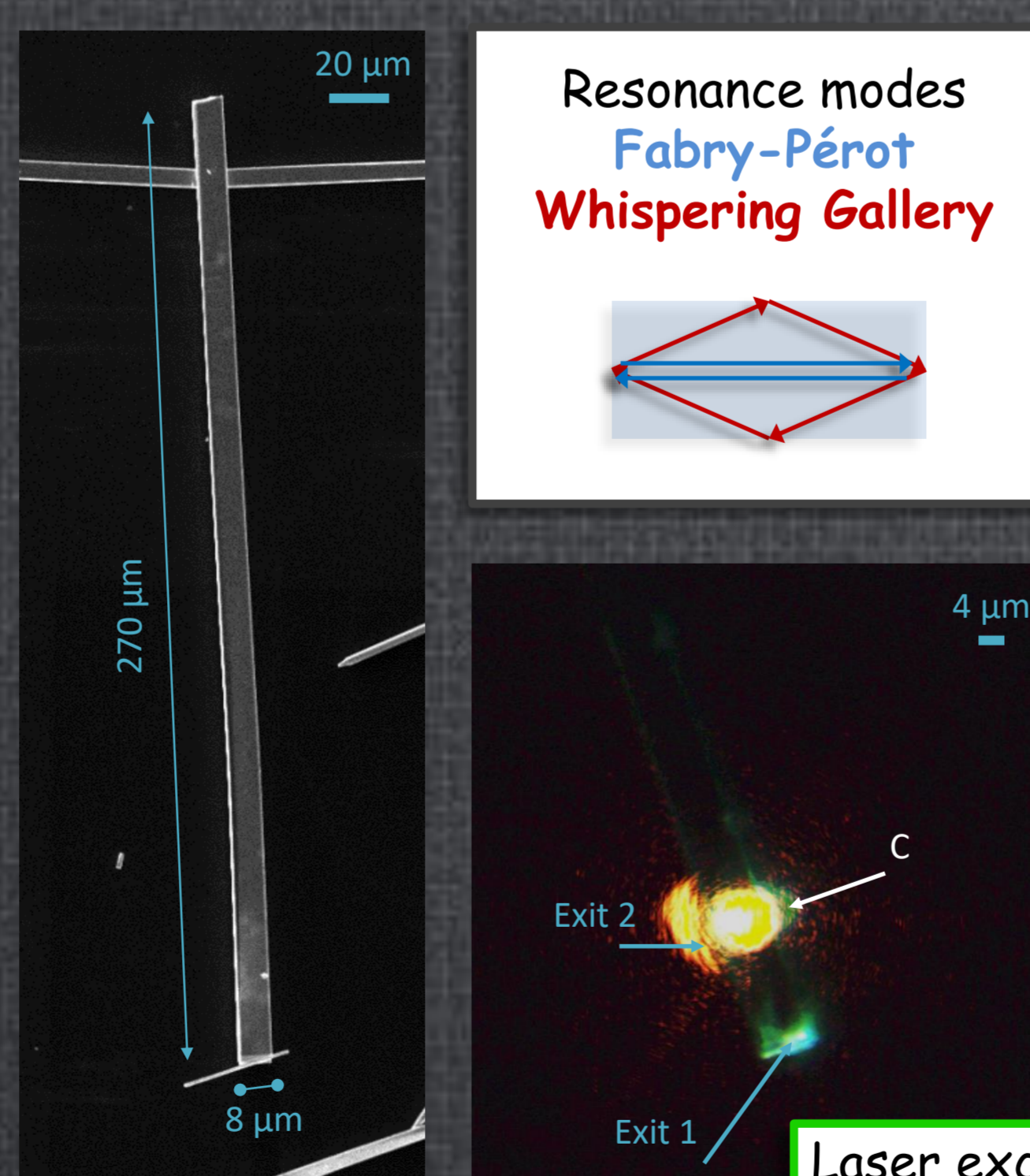


NIR CATHODOLUMINESCENT EMISSION

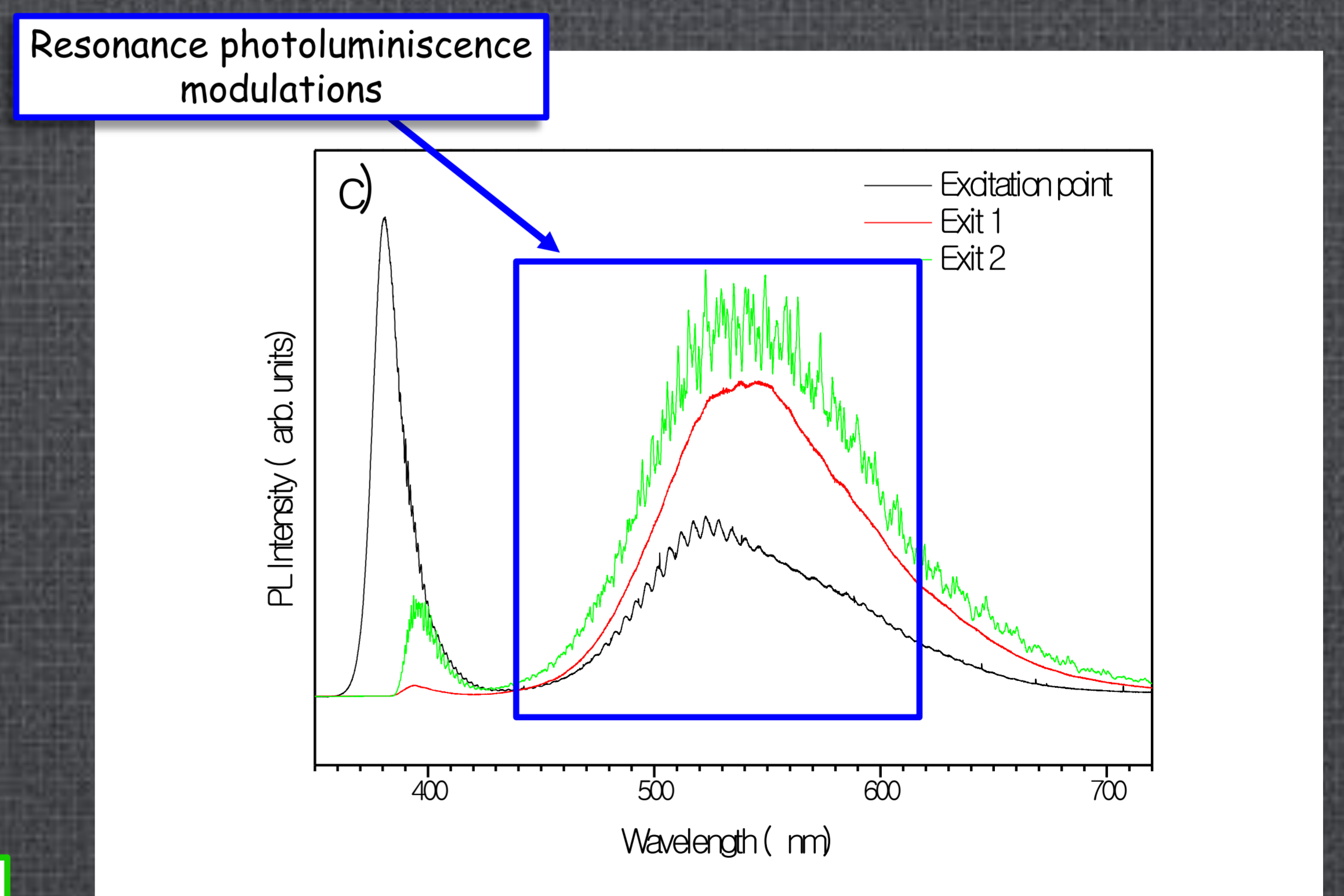


Light guiding

LIGHT GUIDING STRUCTURE EXAMPLE



PHOTOLUMINESCENCE SPECTRA



Conclusions

- ✓ Er and Li doped ZnO nanowires have been grown by vapor-solid method. A large amount of nanostructures from ribbons to wires can be obtained in a very short time.
- ✓ The morphology of the structures changes depending on the dopant atom. Li addition allows a much better control of the obtained structures and 5% wt Li gets uniform in morphology.
- ✓ Luminescence emission changes in the doped structures. Much clearer 1.54 μm (0.8 eV) emission is obtained for 5% wt Li.
- ✓ Light guiding properties are tested and resonance modes are observed in photoluminescence measurement.